

# Condensation of $\alpha$ -cyanocinnamionitriles with 6-bromo-2-naphthol: synthesis of pyrano [2,3-*d*]pyrimidine and pyrano[3,2-*e*] [1,2,4]triazolo[2,3-*c*]pyrimidine derivatives<sup>†</sup>

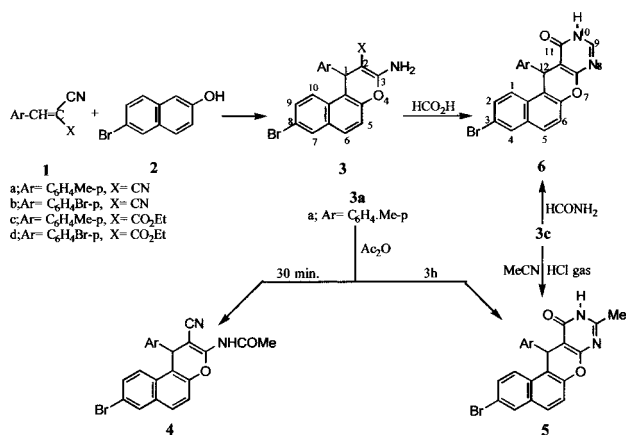
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Naphthopyrans are synthesized by the reaction of cinnamionitriles with 6-bromo-2-naphthol; polysubstituted naphthopyrimidines and naphthopyranotriazolopyrimidines are also prepared.

The considerable biological and medicinal activity of fused 4*H*-pyran has stimulated much research in this field.<sup>1–3</sup> In continuation of our previous work<sup>4–6</sup> on the synthesis of fused pyrans using enamionitriles, we report here the synthesis of a variety of new heterocyclic compounds. Thus, condensation of various substituted  $\alpha$ -cyanocinnamionitriles **1a–d** with 6-bromo-2-naphthol **2** in ethanolic piperidine afford 1:1 adducts.<sup>4–8</sup> Structure **3** (Scheme 1) was established on the basis of the <sup>1</sup>H NMR spectra, which showed 1-H at  $\delta$  5.38 (**3a**) and at  $\delta$  5.48 ppm (**3d**). The increased chemical shift for this signal, compared to the expected value ( $\delta$  4.0–5.0 ppm) for such protons, can be attributed to the deshielding effect of the diamagnetic current of the naphthyl, aryl and allylic  $\pi$ -electrons.<sup>8–10</sup> The UV spectrum of **3b** and **3d** revealed a weak shoulder,<sup>6,11</sup> characteristic for 4*H*-pyran, at  $\lambda_{\text{max}}$  (CH<sub>3</sub>COCH<sub>3</sub>) 275 (log  $\epsilon$  2.84) and 275 nm (log  $\epsilon$  2.83) respectively.



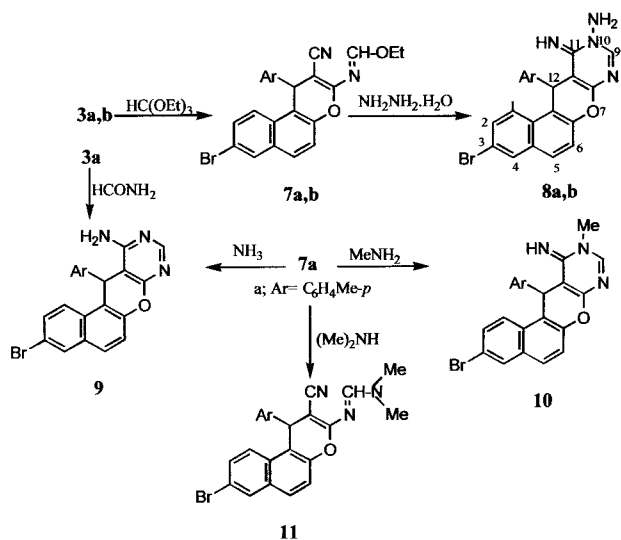
Scheme 1

Interaction of 3-amino-8-bromo-1-(*p*-tolyl)-1*H*-naphtho[2,1-*b*]pyran-2-carbonitrile **3a** with acetic anhydride for 30 min afforded the *N*-acetyl derivative **4**, while heating of **3a** with acetic anhydride under reflux for 3h afforded the naphthopyranopyrimidin-11-one derivative **5**. Structure **5** is supported by an independent synthesis of the same product from **3c** and acetonitrile in the presence of HCl gas<sup>12</sup> (Scheme 1). Treatment of **3a** with formic acid gave the naphthopyrimidin-11-one derivative **6**. The structure of **6** was supported by an independent synthesis from **3c** and formamide (Scheme 1). Structures **4–6** were established by spectral data and analogy with our previous work.<sup>4,6</sup>

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<sup>†</sup> This is a Short Paper, there is therefore no corresponding material in *J. Chem. Research (M)*.

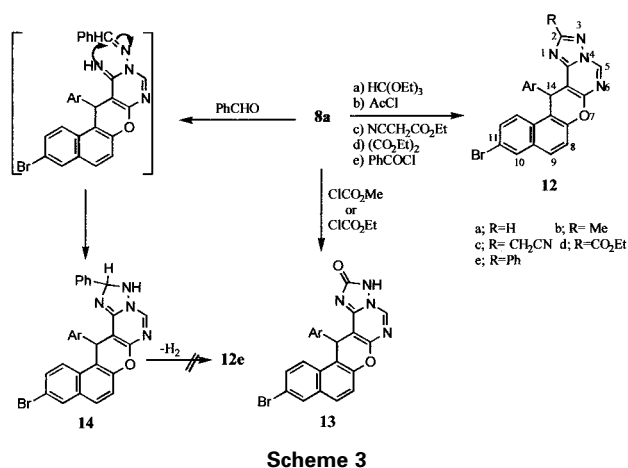
Treatment of **3a,b** with triethyl orthoformate in acetic anhydride at reflux gave the corresponding ethoxymethyleneamino derivatives **7a,b** (Scheme 2). Hydrazinolysis of **7a,b** in ethanol at room temperature afforded the imino derivatives **8a,b** (Scheme 2). Ammonolysis of **7a** in methanol at room temperature afforded the pyrimidine derivative **9**, the structure of which was supported by its independent synthesis from **3a** and formamide (Scheme 2). Reaction of **7a** with methylamine yielded the pyrimidine derivative **10**, while with dimethylamine the open-chain product **11** was obtained (Scheme 2).



Scheme 2

Interaction of **8a** with triethyl orthoformate afforded 11-bromo-14-(*p*-tolyl)-14*H*-naphtho-[1',2':5,6]pyrano[3,2-*e*][1,2,4]triazolo[2,3-*c*]pyrimidine **12a** (Scheme 3), while with acetyl chloride and ethyl cyanoacetate at reflux the corresponding 2-methyl **12b** and 2-acetonitrile **12c** derivative respectively were formed. Reaction of **8a** with diethyl oxalate and benzoyl chloride at reflux afforded the corresponding 2-carboxylate **12d** and 2-phenyl **12e** derivatives respectively, while methyl or ethyl chloroformate in dry benzene afforded the 1:1 adduct **13** (Scheme 3). Structures **7–13** were established by spectral data and analogy with our previous work.<sup>4–6</sup>

Instead of the expected formation of the triazolopyrimidine derivative<sup>13</sup> **12e**, the reaction of **8a** with benzaldehyde in dioxan/piperidine gave the dihydrotriazolopyrimidine derivative **14** (Scheme 3). The proposed structure for **14** was supported by TLC and spectral data.



Scheme 3

## Experimental

Mps are uncorrected. Elemental analyses were carried out in the Microanalytical Laboratories of the Faculty of Science, Cairo University. IR spectra (KBr) were measured on a FT IR/5300 spectrometer. Ultraviolet spectra were recorded on Perkin Elmer Lambda-3B UV-visible spectrophotometer.  $^1\text{H}$  NMR spectra on Varian Mercury (300 MHz) spectrometer and mass spectra on a Shimadzu GC-MS-QP 1000 EX spectrometer.

**Reaction of 1a-d with 6-bromo-2-naphthol:** A solution of **1** (0.01 mol) in ethanol (30 ml) was treated with 6-bromo-2-naphthol **2** (0.01 mol) and piperidine (0.5 ml). The reaction mixture was heated until complete precipitation (reaction times: 15 min. for **1a,b**; 120 min for **1c,d**). The solid product which formed was collected by filtration and recrystallized from a suitable solvent to give **3a-d** (70-80% yield) (Table 1). **3a**:  $\nu_{\text{max}}/\text{cm}^{-1}$  3325, 3283 ( $\text{NH}_2$ ), 2924, 2851 (CH stretching), 2176 (CN);  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 7.25-7.85 (9H,m, arom.), 6.89 (2H,br,  $\text{NH}_2$ ), 5.38 (1H,s, pyran CH), 2.34 (3H,s,  $\text{CH}_3$ ). **3b**:  $\nu_{\text{max}}/\text{cm}^{-1}$  3468, 3319 ( $\text{NH}_2$ ), 2193 (CN). **3d**:  $\nu_{\text{max}}/\text{cm}^{-1}$  3485, 3325 ( $\text{NH}_2$ ), 2980, 2936, 2891 (CH stretching), 1684 (CO ester);  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 7.22-8.20 (9H,m, arom.), 7.19 (2H,br,  $\text{NH}_2$ ), 5.48 (1H,s, pyran CH), 4.10 (2H,q,  $\text{CH}_2$ ,  $J=7.2$  Hz) and 1.26 (3H,t,  $\text{CH}_3$ ,  $J=7.2$  Hz).

**2-Acetylaminio-7-bromo-4-(p-tolyl)-4H-naphtho[2,1-b]pyran-3-carbonitrile (4):** A solution of **3a** (0.01 mol) in acetic anhydride (20 ml) was heated under reflux for 15 min to give the N-acetyl derivatives **4** (81% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3200 (NH), 3047, 2927 (CH stretching), 2206 (CN), 1707 (CO acetyl);  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 11.15 (1H,br, NH), 7.13-8.26 (9H,m, arom.), 5.59 (1H,s, pyran CH), 2.52 (3H,s,  $\text{COCH}_3$ ) and 2.23 (3H,s,  $\text{CH}_3$ ).

**3-Bromo-9-methyl-12-(p-tolyl)-10,11-dihydro-12H-naphtho[1',2':5,6]pyrano[2,3-d]pyrimidin-11-one (5):** (a) A solution of **3a** (0.01 mol) in acetic anhydride (20 ml) was heated under reflux for 3h to give **5** (86% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3260 (NH), 3001, 2850 (CH stretching) and 1651 (CO).

(b) A stream of dry HCl gas was passed through a mixture of **3c** (0.01 mol) and acetonitrile (30 ml) for 4-6h. The reaction mixture was poured into ice-water and basified with 10% ammonium hydroxide solution to give **5** (68% yield) (Table 1).

**3-Bromo-12-(p-tolyl)-10,11-dihydro-12H-naphtho[1',2':5,6]pyrano[2,3-d]pyrimidin-11-one (6):** (a) A solution of **3a** (0.01 mol) in formic acid (20 ml) was heated under reflux for 6h. to give **6** (67% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3400 (NH), 3035, 2995 (CH stretching) and 1675 (CO).

(b) A solution of **3c** (0.01 mol) in formamide (20 ml) was heated under reflux for 6h. to give **6** (73% yield) (Table 1).

**4-Aryl-7-bromo-2-ethoxymethylideneamino-4H-naphtho[2,1-b]pyran-3-carbonitrile (7a,b):** A mixture of **3a,b** (0.01 mol), triethyl orthoformate (0.01 mol) and acetic anhydride (20 ml) was refluxed for 5h to give **7a,b** (75-82% yield) (Table 1). **7a**:  $\nu_{\text{max}}/\text{cm}^{-1}$  2980, 2922, 2860 (CH stretching), 2207 (CN);  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 8.71 (1H,s, CH), 7.11-8.21(9H,m, arom.), 5.52(1H,s, pyran CH), 4.33(2H,q,  $\text{CH}_2$ ,  $J=6.9$  Hz), 2.21 (3H,s,  $\text{CH}_3$ ), 1.32 (3H,t,  $\text{CH}_3$ ,  $J=6.9$  Hz). **7b**:  $\nu_{\text{max}}/\text{cm}^{-1}$  2986, 2937, 2855 (CH stretching), 2206 (CN);  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 8.73 (1H,s, CH), 7.22-8.23(9H,m, arom.), 5.65 (1H,s, pyran CH), 4.34 (2H,q,  $\text{CH}_2$ ,  $J=6.9$  Hz) and 1.32 (3H,t,  $\text{CH}_3$ ,  $J=6.9$  Hz).

**10-Amino-12-aryl-3-bromo-11-imino-10,11-dihydro-12H-naphtho[1',2':5,6]pyrano[2,3-d]pyrimidine (8a,b):** A solution of **7a,b** (0.01

**Table 1** Characterization data for newly synthesized compounds

Compound No.	Mp (T/°C) <sup>a</sup>	Molecular formula	Found (required) (%)	
			C	H
<b>3a</b>	215 <sup>b</sup>	C <sub>21</sub> H <sub>15</sub> BrN <sub>2</sub> O	64.4 (64.62)	3.6 (3.85)
<b>3b</b>	250 <sup>b</sup>	C <sub>20</sub> H <sub>12</sub> Br <sub>2</sub> N <sub>2</sub> O	52.7 (52.86)	2.5 (2.64)
<b>3c</b>	185 <sup>b</sup>	C <sub>23</sub> H <sub>20</sub> BrN <sub>2</sub> O <sub>3</sub>	63.3 (63.16)	4.4 (4.58)
<b>3d</b>	165 <sup>b</sup>	C <sub>22</sub> H <sub>17</sub> Br <sub>2</sub> N <sub>2</sub> O <sub>2</sub>	52.8 (52.69)	3.5 (3.39)
<b>4</b>	255	C <sub>23</sub> H <sub>17</sub> BrN <sub>2</sub> O <sub>2</sub>	63.7 (63.89)	3.8 (3.94)
<b>5</b>	310	C <sub>23</sub> H <sub>17</sub> BrN <sub>2</sub> O <sub>2</sub>	63.9 (63.89)	4.0 (3.94)
<b>6</b>	140 <sup>b</sup>	C <sub>22</sub> H <sub>15</sub> BrN <sub>2</sub> O <sub>2</sub>	63.0 (63.16)	3.3 (3.59)
<b>7a</b>	178	C <sub>24</sub> H <sub>19</sub> BrN <sub>2</sub> O <sub>2</sub>	64.7 (64.57)	4.4 (4.26)
<b>7b</b>	190	C <sub>23</sub> H <sub>16</sub> Br <sub>2</sub> N <sub>2</sub> O <sub>2</sub>	51.1 (51.30)	2.7 (2.97)
<b>8a</b>	250	C <sub>22</sub> H <sub>17</sub> BrN <sub>4</sub> O	61.0 (61.11)	3.8 (3.94)
<b>8b</b>	265	C <sub>21</sub> H <sub>14</sub> Br <sub>2</sub> N <sub>4</sub> O	50.9 (50.81)	3.0 (2.82)
<b>9</b>	297	C <sub>22</sub> H <sub>16</sub> BrN <sub>3</sub> O	63.2 (63.31)	3.6 (3.84)
<b>10</b>	278	C <sub>23</sub> H <sub>18</sub> BrN <sub>3</sub> O	64.0 (64.04)	4.0 (4.18)
<b>11</b>	248	C <sub>24</sub> H <sub>20</sub> BrN <sub>3</sub> O	64.9 (64.72)	4.5 (4.49)
<b>12a</b>	310 <sup>c</sup>	C <sub>23</sub> H <sub>15</sub> BrN <sub>4</sub> O	62.4 (62.44)	3.4 (3.39)
<b>12b</b>	285 <sup>c</sup>	C <sub>24</sub> H <sub>15</sub> BrN <sub>4</sub> O	63.0 (63.16)	3.6 (3.73)
<b>12c</b>	287	C <sub>25</sub> H <sub>16</sub> BrN <sub>5</sub> O	62.3 (62.37)	3.4 (3.33)
<b>12d</b>	215	C <sub>26</sub> H <sub>19</sub> BrN <sub>4</sub> O <sub>3</sub>	60.6 (60.70)	3.7 (3.70)
<b>12e</b>	290 <sup>d</sup>	C <sub>29</sub> H <sub>19</sub> BrN <sub>4</sub> O	67.2 (67.18)	3.7 (3.67)
<b>13</b>	292 <sup>c</sup>	C <sub>23</sub> H <sub>15</sub> BrN <sub>4</sub> O <sub>2</sub>	60.1 (60.26)	3.1 (3.28)
<b>14</b>	295 <sup>d</sup>	C <sub>29</sub> H <sub>21</sub> BrN <sub>4</sub> O	67.0 (66.92)	4.1 (4.04)

<sup>a</sup>From benzene unless indicated otherwise. <sup>b</sup>From ethanol.

<sup>c</sup>From DMF <sup>d</sup>From dioxan

mol) and hydrazine hydrate (99%, 5 ml) in ethanol (50 ml) was stirred for 45 min to give **8a,b** (85-87% yield) (Table 1). **8a**:  $\nu_{\text{max}}/\text{cm}^{-1}$  3375, 3300 ( $\text{NH}_2$ ), 3161 (NH);  $m/z$  434/432 ( $\text{M}^+$ , 25/26%), 418/416 (99/100), 342/340 (15/17), 272/270 (4/4), 246/244 (5/6), 165 (14), 111 (7), 73 (9). **8b**:  $\nu_{\text{max}}/\text{cm}^{-1}$  3354, 3325 ( $\text{NH}_2$ ) and 3168 (NH).

**11-Amino-3-bromo-12-(p-tolyl)-12H-naphtho[1',2':5,6]pyrano[2,3-d]pyrimidine (9):** (a) Compound **9** was prepared from **7a** (0.01 mol) and  $\text{NH}_3$  gas according to the procedure described for **8**; (88% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3431, 3319 ( $\text{NH}_2$ ) and 1649 (C=N).

(b) Compound **9** was prepared from **3a** (0.01 mol) and formamide (0.01 mol) according to the procedure described for **6** (method b) (65% yield) (Table 1).

**3-Bromo-10-methyl-11-imino-12-(p-tolyl)-10,11-dihydro-12H-naphtho[1',2':5,6]pyrano[2,3-d]pyrimidine (10):** Compound **10** was prepared from **7a** (0.01 mol) and methylamine (0.01 mol) according to the procedure described for **8** (83% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3342 (NH), 3020, 2918, 2876 (CH stretching), 1645 (C=N),  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 8.21 (1H,s, pyrimidine CH), 7.03-8.10 (9H,m, arom.), 7.00 (1H,br, NH), 5.88 (1H,s, pyran CH), 3.35 (3H,s, N- $\text{CH}_3$ ) and 2.15 (3H,s,  $\text{CH}_3$ ).

**7-Bromo-2-dimethylaminomethylideneamino-4-(p-tolyl)-4H-naphtho[2,1-b]pyran-3-carbonitrile (11):** Compound **11** was prepared from **7a** (0.01 mol) and dimethylamine (0.01 mol) according to the procedure described for **8** (87% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  2920, 2855, 2814 (CH stretching), 2195 (CN),  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 8.48 (1H,s, =CH), 7.01-8.18 (9H,m, arom.), 5.39 (1H,s, pyran CH), 3.17 (3H,s,  $\text{NCH}_3$ ), 3.01 (3H,s,  $\text{NCH}_3$ ) and 2.22 (3H,s,  $\text{CH}_3$ ).

**11-Bromo-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-e][1,2,4]triazolo[2,3-c]pyrimidine (12a):** A solution of **8a** (0.01 mol) and triethyl orthoformate (0.01 mol) in dry benzene was refluxed for 6h to give **12a** (79% yield) (Table 1),  $m/z$  444/442 ( $\text{M}^+$ , 23/21%), 353/351 (95/100), 299/297 (6/6), 218 (7), 192 (1), 165 (3), 112 (7) and 77 (9).

**11-Bromo-2-methyl-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-e][1,2,4]triazolo[2,3-c]pyrimidine (12b):** Compound **12b** was prepared from **8a** (0.01 mol) and acetyl chloride (0.01 mol) according to the procedure described for **12a** (84% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3060, 2980, 2950 (CH-stretching), 1643 (C=N),  $\delta_{\text{H}}$  ( $[\text{}^2\text{H}_6]\text{DMSO}$ ) 8.74 (1H,s, pyrimidine CH), 7.06-8.20 (9H,m, arom.), 6.49 (1H,s, pyran CH), 3.57(3H,s, triazolo  $\text{CH}_2$ ) and 2.16 (3H,s,  $\text{CH}_3$ ).

**11-Bromo-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-e][1,2,4]triazolo[2,3-c]pyrimidin-2-acetonitrile (12c):** A mixture of **8a** (0.01 mol) and ethyl cyanoacetate (0.01 mol) and absolute ethanol (20 ml) was refluxed for 6h to give **12c** (65% yield) (Table 1),  $\nu_{\text{max}}/\text{cm}^{-1}$  3059, 3030, 2930, 2876 (CH stretching) and 2255 (CN).

*Ethyl 11-bromo-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-*

*e*][1,2,4]triazolo[2,3-*c*]pyrimidine-2-carboxylate (**12d**): Compound **12d** was prepared from **8a** (0.01 mol) and ethyl oxalate (0.01 mol) according to the procedure described for **12c**, (80% yield) (Table 1),  $\nu_{\max}/\text{cm}^{-1}$  3111, 2990, 2891 (CH stretching), 1718 (CO), 1618 (C=N),  $\delta_{\text{H}}$  ( $^2\text{H}_6$ ]DMSO) 8.74 (1H,s, pyrimidine CH), 7.06–8.31 (3H,m, arom.), 6.33 (1H,s, pyran CH), 3.99 (2H,q,CH<sub>2</sub>,  $J=6.9$  Hz), 2.17 (3H,s,CH<sub>3</sub>), 1.10 (3H,t,CH<sub>3</sub>,  $J=6.9$  Hz),  $m/z$  516/514 (M<sup>+</sup>, 1/1%), 420/418 (49/54), 328/326 (86/100), 193 (16), 166 (1), 140 (2), 112 (3) and 65 (11).

*11-Bromo-2-phenyl-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-*e*][1,2,4]triazolo[2,3-*c*]pyrimidine* (**12e**): Compound **12e** was prepared from **8a** (0.01 mol) and benzoyl chloride (0.01 mol) according to the procedure described for **8a**, (62% yield) (Table 1),  $\nu_{\max}/\text{cm}^{-1}$  3005 (CH-stretching), 1633 (C=N),  $\delta_{\text{H}}$  ( $^2\text{H}_6$ ]DMSO) 9.67 (1H,s, pyrimidine CH), 7.03–8.31 (14H,m, arom.), 6.37 (1H,s, pyran CH) and 2.13 (3H,s, CH<sub>3</sub>).

*11-Bromo-2,3-dihydro-2-oxo-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-*e*][1,2,4]triazolo-*[2,3-*c*]pyrimidine** (**13**): Compound **13** was prepared from **8a** (0.01 mol) and methyl chloroformate or ethyl chloroformate (0.01 mol) according to the procedure described for **12a**, (74% yield) (Table 1),  $\nu_{\max}/\text{cm}^{-1}$  3362 (NH), 2049, 2989, 2856 (CH stretching) and 1647 (CO).

*11-Bromo-2,3-dihydro-2-phenyl-14-(p-tolyl)-14H-naphtho[1',2':5,6]pyrano[3,2-*e*][1,2,4]triazolo-*[2,3-*c*]pyrimidine** (**14**): A mixture of **8a** (0.01 mol), benzaldehyde (0.01 mol), dioxan (20 ml) and piperidine (0.5 ml) was refluxed for 16h to give **14** (85% yield) (Table 1),  $\nu_{\max}/\text{cm}^{-1}$  3188 (NH), 3028, 2920, 2891, 2855 (CH stretching),  $\delta_{\text{H}}$  ( $^2\text{H}_6$ ]DMSO) 11.17 (1H, br, NH), 8.39 (1H,s, pyrimidine CH), 7.00–8.29 (14H,m, arom.), 6.68 (1H,s, pyran CH), 3.58 (1H,s, triazoline CH), 2.12 (3H,s, CH<sub>3</sub>);  $m/z$  522/520 (M<sup>+</sup>, 13/12%), 418/416 (78/100), 391/389 (15/19), 300/298 (13/14), 218 (5), 191 (1), 164 (7), 135 (4) and 72 (2).

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